

LANSDOWNE SCHOOL NEWSLETTER

I would like to congratulate all the students who have been out on work experience over the last two weeks. There have been some wonderful placements at sports centres, marketing agencies, and retail outlets. The students have really enjoyed their time in the world of work and have been able to reflect on their weeks in their work experience diaries. Special thanks to Mr Anatoliotis for arranging the placements and to Ricky and Valerie for assisting the students with travelling to work places. We see work experience as an important part of the development of our young people. We are very grateful to all the businesses and organizations that support us with this.

Mr Jukes



Class of the Week

12RM

Congratulations!!!

Work experience

Year 11 and 12 students have been on work experience since last week. In collaboration with local businesses in various sectors we provide our students with the opportunity to experience a working environment that they selected for a whole week. We manage to offer either “in-house” or business placements to all our Year 11, 12 and 13 students every year. We strongly value that students gain important insights into their career interests and values, which will help them decide about their future. Career management skills, such as organizing, negotiating, networking and self-presentation can be practised. The feedback from the students, parents and employers is very positive and the connection with the local community is building from strong to stronger.



Dates for your Diary

Bake Sale

Friday 14th February



Don't forget we have one of our famous Lansdowne Bake Sales next Friday 10-11am. Our theme is “Food around the world” and proceeds go to the Brixton Soup Kitchen.

what we learned this week

Science

In Science we have learnt about separation techniques by making Copper Sulphate crystals. We had to use filtration and evaporation.

Making Crystals

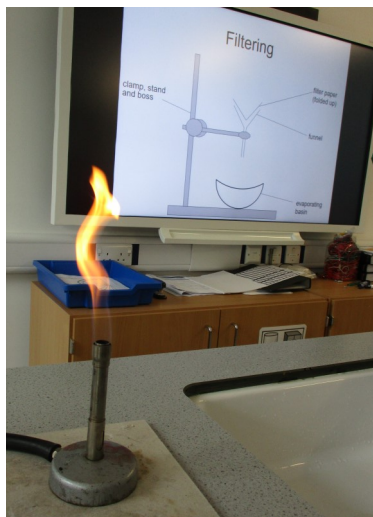


You are going to make some crystals of copper sulphate. All you have to do this is some black powder called copper oxide and some sulphuric acid. You can also use any equipment in the lab.

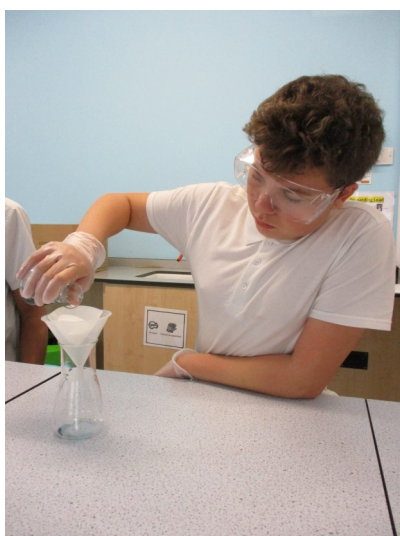
Step 1

Put 25ml of sulphuric acid into a conical flask. Add a spatula of copper oxide and swirl it about. Keep adding the copper oxide until no more can dissolve and it lies on the bottom of the conical flask.

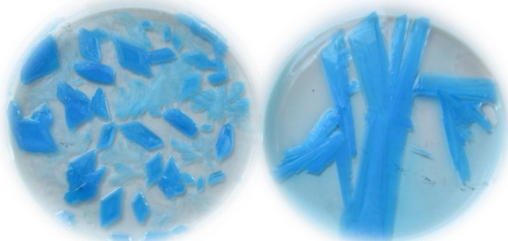
Observations: What happened? Was there a chemical reaction happening?



Before



Two days later



After School Clubs

Tuesdays

- Football
- Yoga
- ICT
- Engine Shed

Thursdays

- Art
- Dodge ball
- Rock Band
- Enterprise
- Games

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